<u>Heisenberg Uncertainty</u> <u>Inequality</u>

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• In 1927, W. Heisenberg (Zeit. Physik 43 (1927), 172 - ; Univ. Chicago Press, 1930 ; and Dover edit., New York, 1949) demonstrated the impossibility to specify simultaneously the position and the momentum of an electron within an atom. The following result named, Heisenberg uncertainty inequality, is not actually due to Heisenberg. In 1928, according to H. Weyl (S. Hirzel, Leipzig, 1928; and Dover edit., New York, 1950) this result is due to W. Pauli.

Heisenberg Uncertainty Inequality: If $f: R \to C$ is a complex valued function of a random real variable *x* such that $f \in L^2(R)$, then the product of the second moment

of the random real x for $|f|^2$ and the second moment of the random real ξ for $|\hat{f}|^2$

is at least $\int_{R} |f(x)|^2 dx / 4\pi$, where \hat{f} is the Fourier transform of f, such that

$$\hat{f}(\xi) = \int_{R} e^{-2i\pi\xi x} f(x) dx \text{ and } f(x) = \int_{R} e^{2i\pi\xi x} \hat{f}(\xi) d\xi \text{ with } i = \sqrt{-1}.$$

According to N. Wiener ("the Fourier integral and certain of its applications", Cambridge, 1933) a pair of transorms cannot both be very small. This inequality plays an important role in different aspects of Fourier and Time – Frequency Analysis. A huge number of well-written books and overview-papers deals with uncertainty relations.